





Study of various configurations of Micro Hollow Cathode Discharges in Ar/N_2 used for boron nitride PECVD

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Joint GEC-IOPS / OLTPD seminar – March 7, 2023

Team in LSPM

• <u>Colleagues</u>:

Xavier Aubert, Ovidiu Brinza, Kristaq Gazeli, Laurent Invernizzi, Guillaume Lombardi, Vianney Mille, Swaminathan Prasanna, Alexandre Tallaire, Ludovic William

- <u>Post-docs</u>:
 - Hiba Kabbara (2018-2019)
 - Manoel Jacquemin (2021-2022)
- <u>PhD students</u>:
 - Salima Kasri (2015-2019)
 - Alice Remigy (2019 2022)
 - Belkacem Menacer (2022)









Collaborations

 Colleagues from LPGP: Gérard Beauville, Michel Fleury, Stéphane Pasquiers, Joao Santos Sousa



• Colleagues from synchrotron SOLEIL: Nelson de Oliveira and Laurent Nahon



• Colleagues from LTM and LIPhy: Nader Sadeghi



 Colleagues from MPQ: Clément Barraud and Maria Luisa Della Rocca



Agenda of the presentation

- I. Introduction
- II. Deposition of boron nitride by PECVD: description of the process and few deposition results
- III. Atomic nitrogen production in the microplasma source
- IV. Atomic nitrogen transport in the deposition chamber
- V. Conclusion and perspectives



My laboratory: LSPM



- Laboratory of Process and Material Sciences; created January 1st, 2011
- Located in France, in the north of Paris





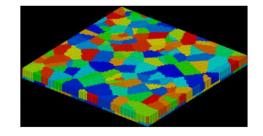


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LSPM: few numbers

- 85 permanent positions:
 - 10 CNRS researchers
 - 55 associated professors/professors
 - 20 engineers/technicians/administrators

• 45 PhD students and post-docs



• 4/5 PhD government grants/year



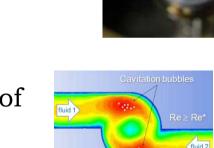


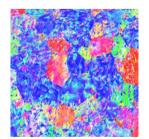
LSPM: scientific axis

- <u>3 scientific axis</u>:
 - 1) Development of elaboration and transformation process of materials

2) Structural caracteristics and properties of materials

3) Integration of materials in systems, devices and process







University Sorbonne Paris Nord





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University Sorbonne Paris Nord

- 5 campus
- 26000 students
- 28 laboratories
- 1500 assistant professors and professors
- 730 people in the administration and technical staff





Deposition of Boron Nitride by microplasmas: process description





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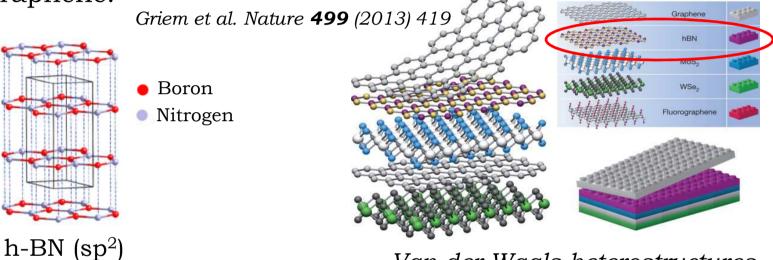
Context

- New research axis since 2016 in LSPM: "Microplasmas for innovative materials".
- <u>Idea</u>: choice of a material and a versatile process
 - → control of the morphological and structural characteristics of the films
- <u>Goals</u>:
 - To develop an efficient new plasma reactor for the deposition of high potential materials on large surfaces, easily adaptable to established technologies
 - To optimize the quality of the deposited layers by understanding the mechanisms involved in growth processes using reactive plasmas
- <u>Target material</u>: hexagonal boron-nitride (h-BN)



Why h-BN?

• Large band gap, thermally and chemically stable, ideal substrate for graphene.

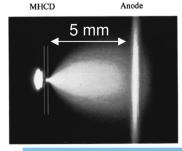


Van der Waals heterostructures

- Strategic material highly demanded for electronic and optoelectronic applications, also quantum technologies.
- But the development of applications based on this material is still held back by the poor availability of high-quality epitaxial films on large areas → new process based on microplasmas.

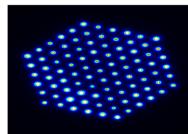


- Low temperature plasma: reduction of the deposition temperature compared to conventional processes.
- Production of high electron density: high degree of dissociation of nitrogen.
- Deposition on large surfaces: MCSD and array of MHCDs.



Ar 160 Torr

Stark et al. PPCF **85** (1999) 2075 Makasheva et al. PPCF **49** (2007) B233

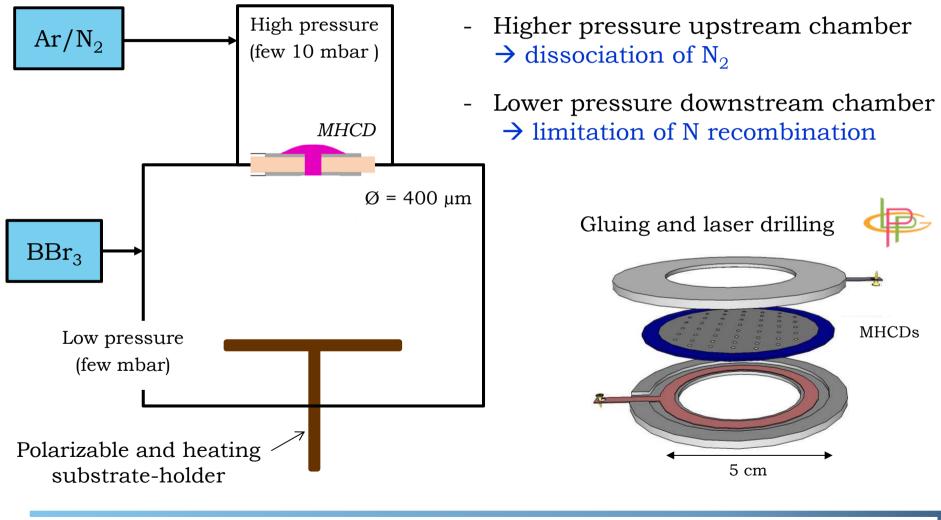


Pulsed power supply

91 MHCDs in pure N_2

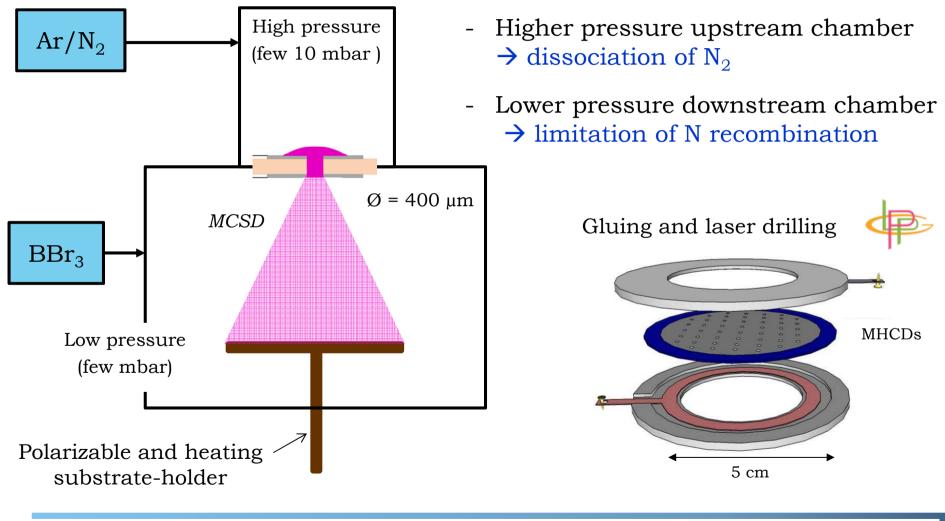


PECVD process



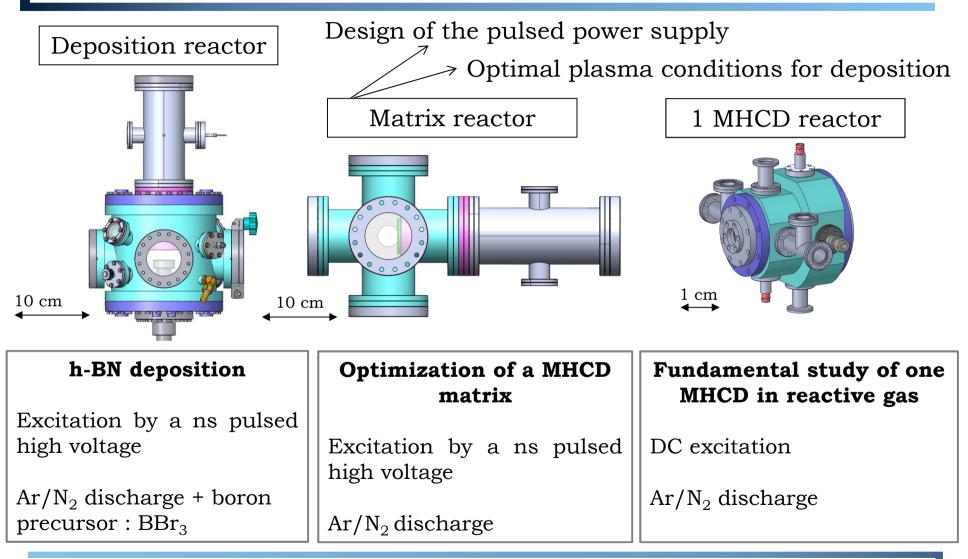


PECVD process



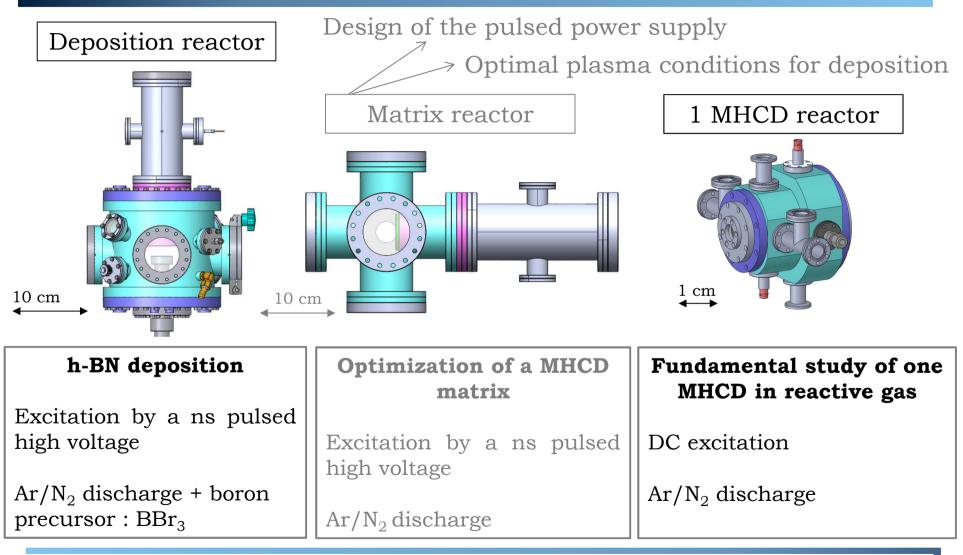


Developed reactors



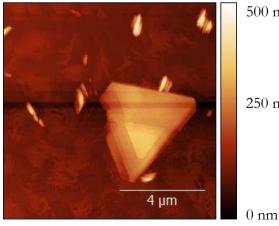


Developed reactors





Deposition reactor: few results



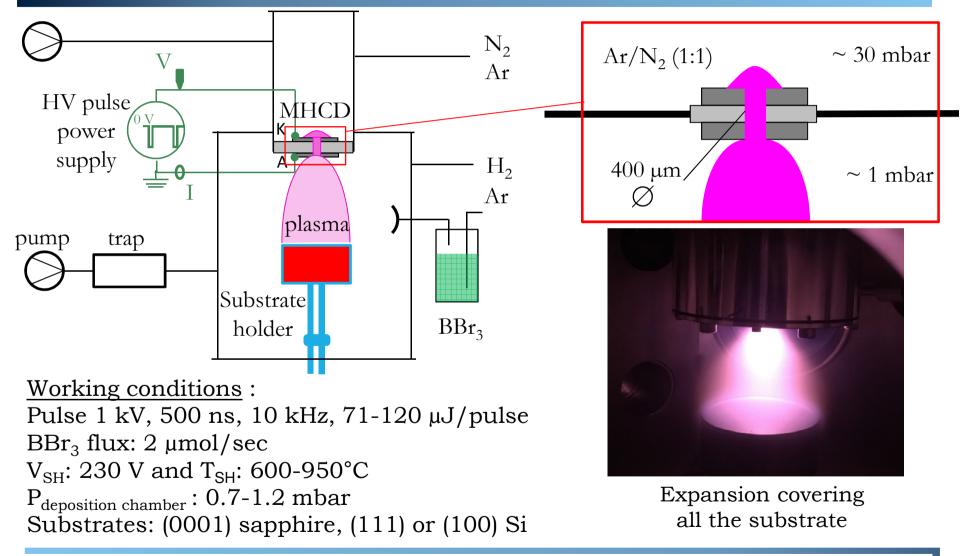
500 nm

250 nm



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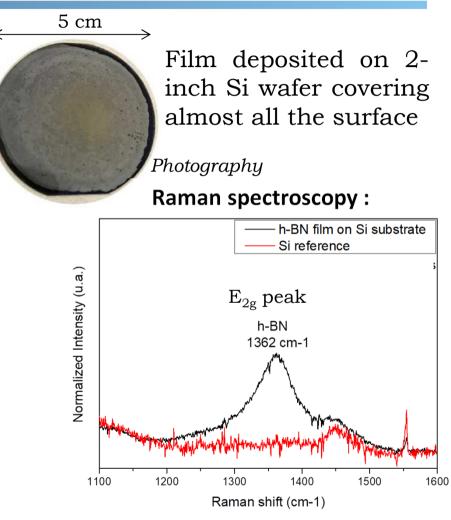
Experimental setup for h-BN deposition

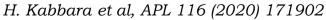




Deposition results

- Deposition on 2-inch silicon (Duration: 2-7h)
- Clear signature of h-BN at 1362 $cm^{-1} \Rightarrow$ Proof of feasibility
- Deposition rate: v ~ 35 nm/h
- Crystalline quality probably moderate considering the broad Raman line at half maximum
- Chemical composition: 55% of B and 45% of N (very close concentration)
- -Effect of the distance MHCD/substrate

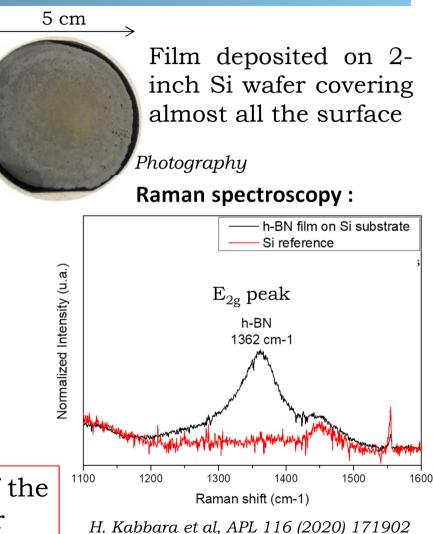






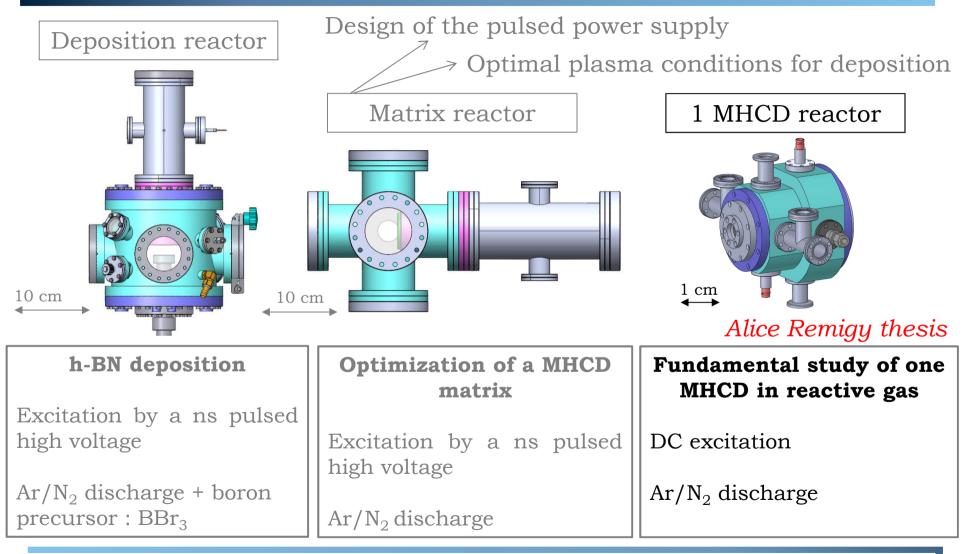
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- Optimisation of the process → study of the discharge: density of N in particular



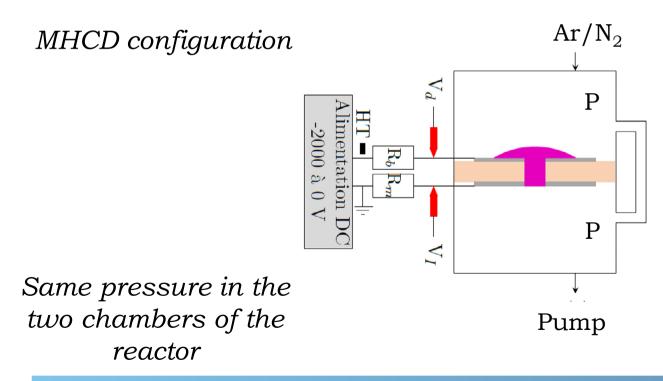


Developed reactors



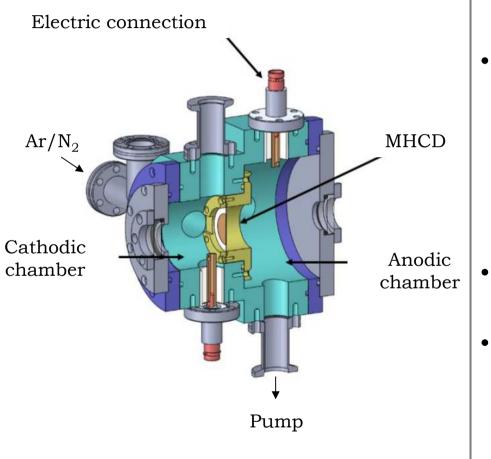


Study of the 1 MHCD reactor: production of atomic nitrogen





Tools



- Electrical characterization → discharge resistance
- VUV Fourier transform absorption spectro. (SOLEIL synchrotron) and TALIF \rightarrow atomic nitrogen density





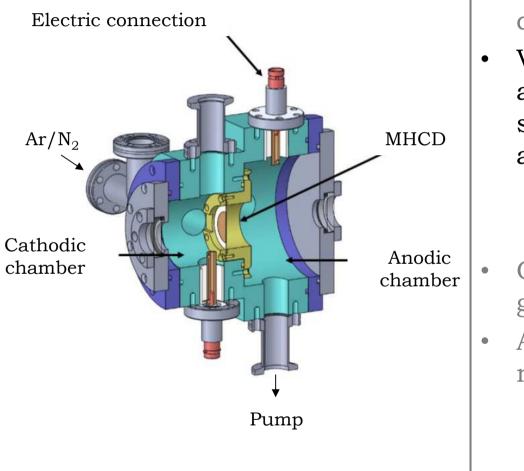
- OES \rightarrow electron density and gas temperature
- Absorption spectroscopy → metastable argon density



• Global model 0D



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- VUV Fourier transform absorption spectro. (SOLEIL synchrotron) and TALIF \rightarrow atomic nitrogen density





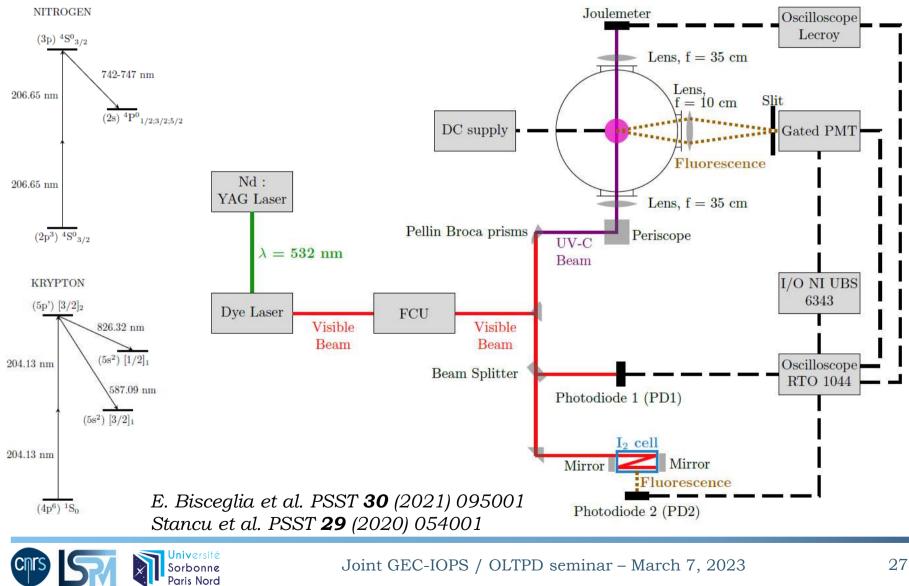
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• Global model 0D

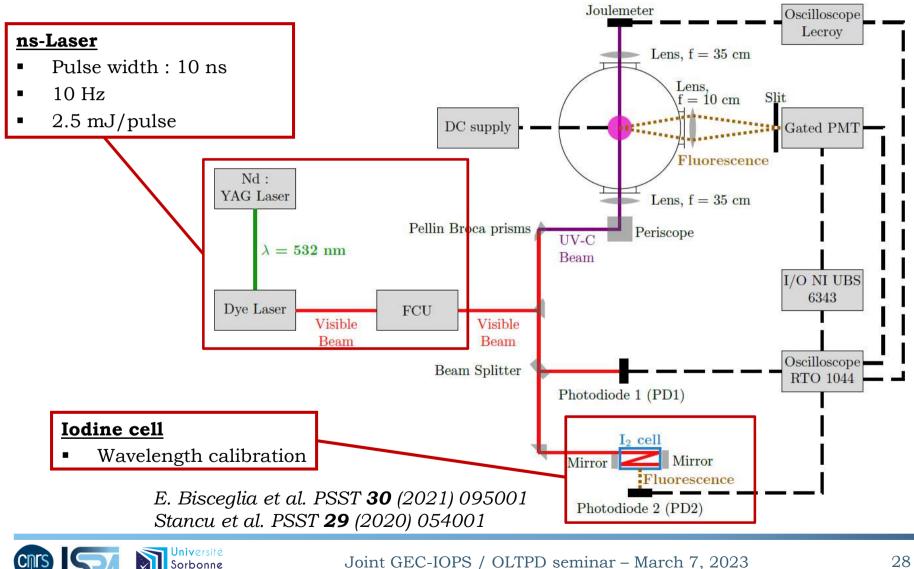


Two photon Absorption Laser Induced Fluorescence (TALIF)



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Two photon Absorption Laser Induced Fluorescence (TALIF)



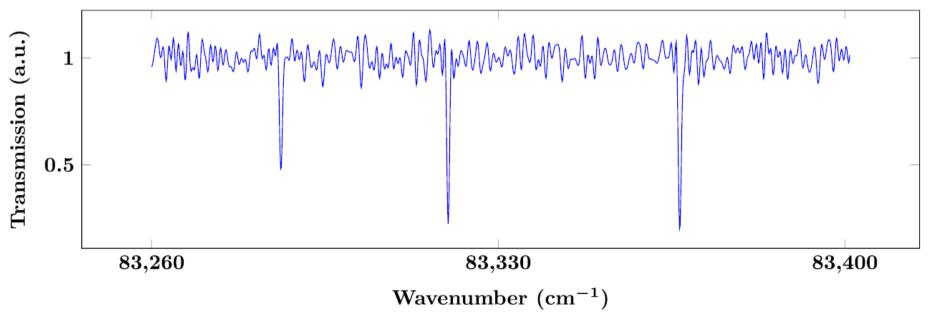
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Fourier transform absorption spectroscopy (FTS VUV)

Direct measurement of N in the ground state studying the transitions in the vacuum UV.
K. Niemi et al. APL **103** (2013) 034102

SYNCHROTRON

• Probed transition (120 nm): N $(2p^3 {}^4S_{3/2}) \rightarrow$ Triplet N $(3s {}^4P_{1/2, 3/2, 5/2})$



Example of a normalized non deconvoluted transmission spectrum



Dumitrache et al. PSST 31 (2022) 015004

Global model

- Volume-averaged (0D) model: temporal variations of volume-averaged quantities (densities and temperature)
- Particle balance:

$$\frac{dn_{\alpha}}{dt} = G_{\alpha} - P_{\alpha}$$

• Electron power balance:

$$\frac{d}{dt}\left(\frac{3}{2}n_e eT_e\right) = P_{\rm abs} - P_{\rm dis}$$



Global model

Power balance:

Modelisation of the cathodic region

$$P_{\rm abs} = \left(\frac{A}{S_C} \kappa I_d V_d \right)$$

Fraction of power absorbed by the electrons:

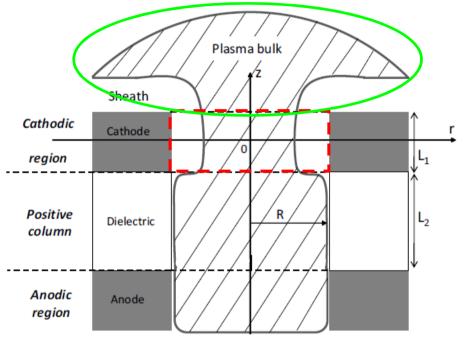
$$\kappa = \frac{\bar{P}_e}{\bar{P}_e + \bar{P}_i} \simeq \frac{1}{1 + \gamma} \frac{1}{\ln(1 + \frac{1}{\gamma})}$$

Particle balance:

Electropositive plasma: Ar-N $_2$ mixture

- 14 species
- 69 volume reactions
- T_g=470 K

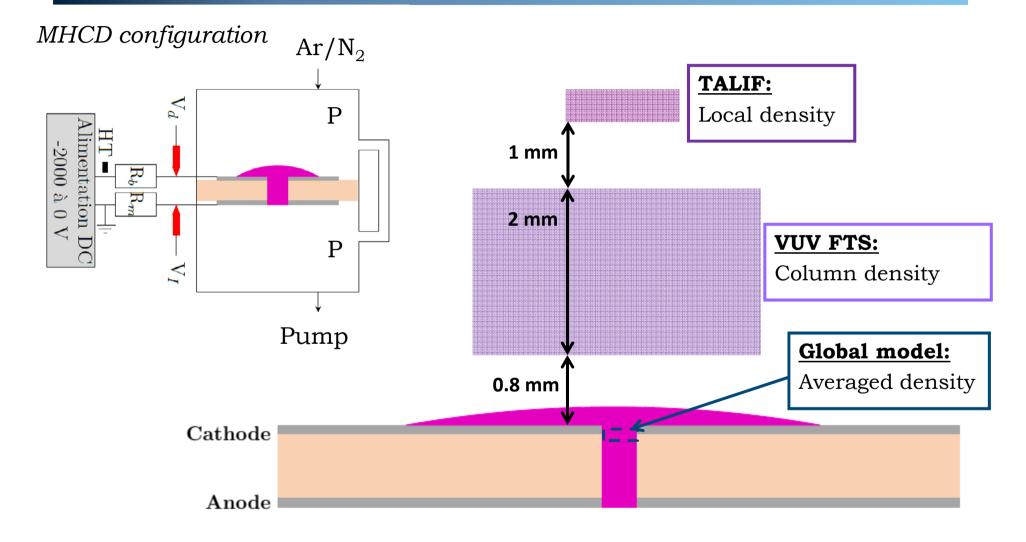




C. Lazzaroni et al. PSST **20** (2011) 055004

14 considered species: Ar, Ar⁺, Ar₂⁺, Ar^m, Ar^r, Ar^{4p}, N₂, N, N₂⁺, N₄⁺, N₂(A), N₂(B), N₂(C) and e⁻

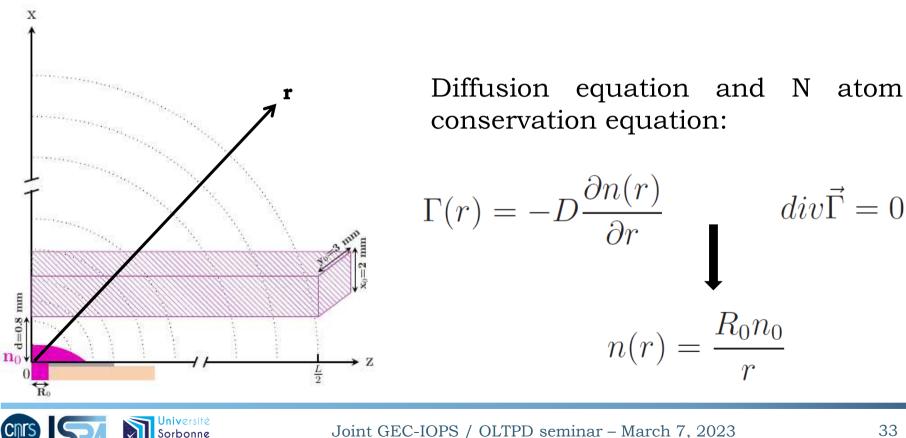
Experiments/model comparison?





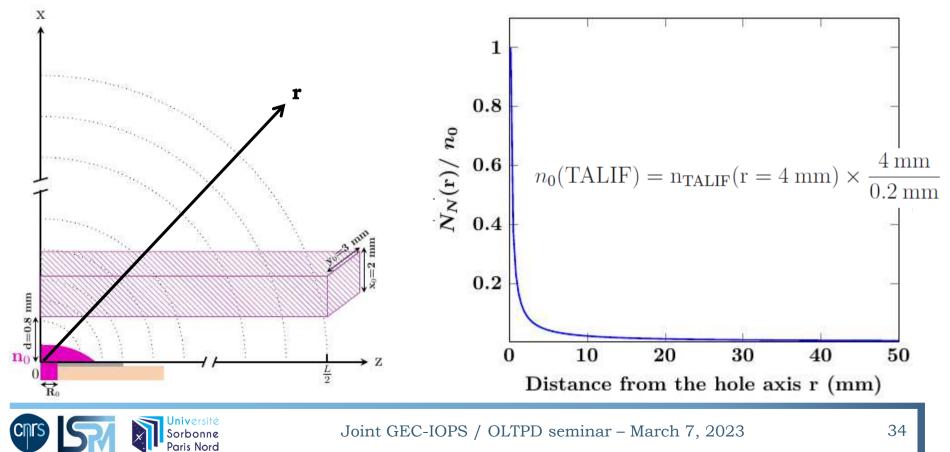
Model of diffusion: n(r)

- \rightarrow VUV FTS measurements: estimation of the absorption length to get the N density from the column density
- \rightarrow Comparison of n_N measurements by FTS and TALIF



Model of diffusion: n(r)

- → VUV FTS measurements: estimation of the absorption length to get the N density from the column density
- $\boldsymbol{\rightarrow}$ Comparison of n_{N} measurements by FTS and TALIF



Model of diffusion: eff. absorption length

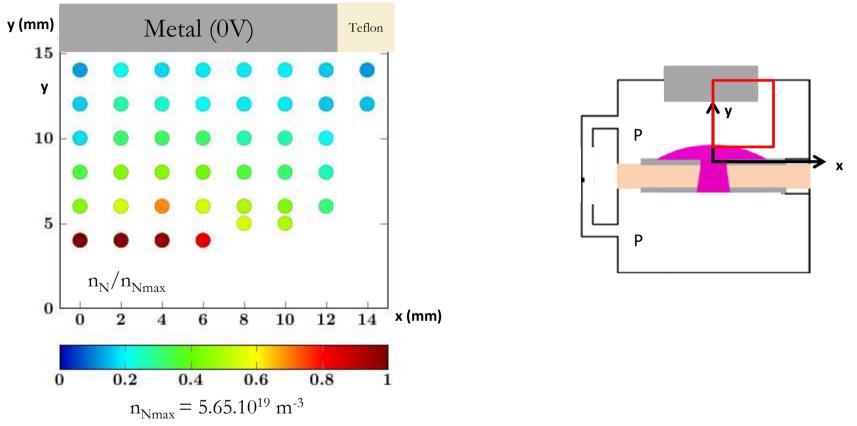
- Density in cartesian coordinates: $n(x, y, z) = \frac{R_0 n_0}{\sqrt{x^2 + y^2 + z^2}}$
- Density profile integrated on the probed volume by the beam \rightarrow effective absorption length $l_{abs}(N)$

$$\begin{aligned} Abs(\sigma) &= \frac{k(\sigma)}{x_0 y_0} \int_d^{d+x_0} \int_{-y_0/2}^{y_0/2} \int_{-L/2}^{L/2} n(x, y, z) dx dy dz \\ Abs(\sigma) &= k(\sigma) n_0 \frac{R_0}{x_0 y_0} \int_d^{d+x_0} \int_{-y_0/2}^{y_0/2} \int_{-L/2}^{L/2} \frac{1}{\sqrt{x^2 + y^2 + z^2}} dx dy dz \\ Abs(\sigma) &= k(\sigma) n_0 l_{abs}(N) \end{aligned}$$

• Numerical integration $\rightarrow l_{abs}(N)=1.58 \text{ mm}$ and $n_{FTS}=n_0=n_N(r=R_0)$

A. Remigy et al. J. Phys. D **55** (2022) 105202

Spatial mapping of N by TALIF

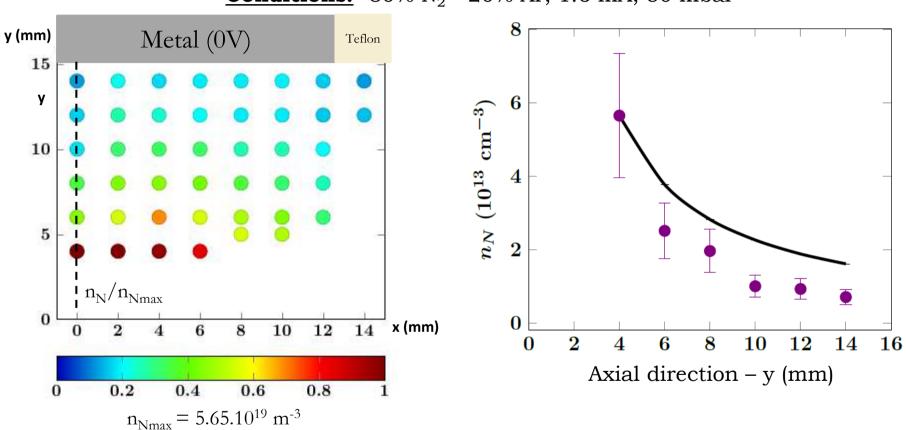


<u>Conditions:</u> 80% N₂ - 20% Ar, 1.6 mA, 50 mbar

• High density near the cathode surface and then fast decay

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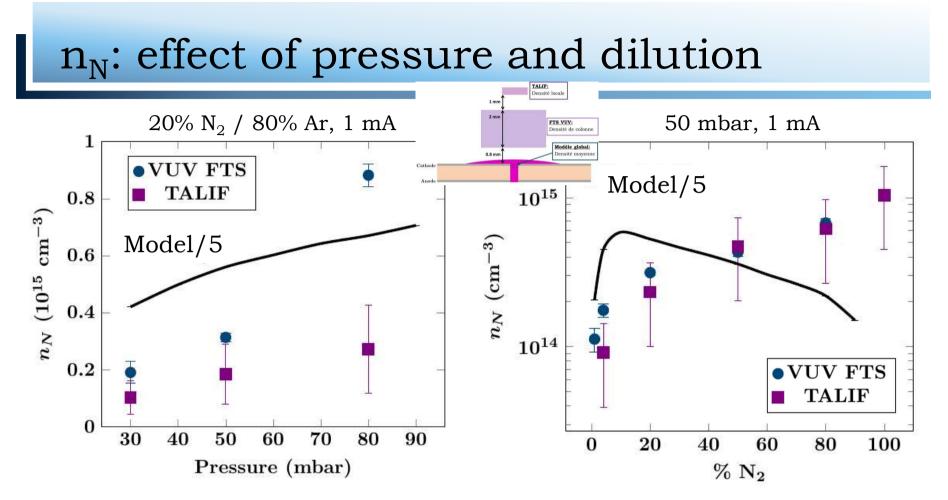
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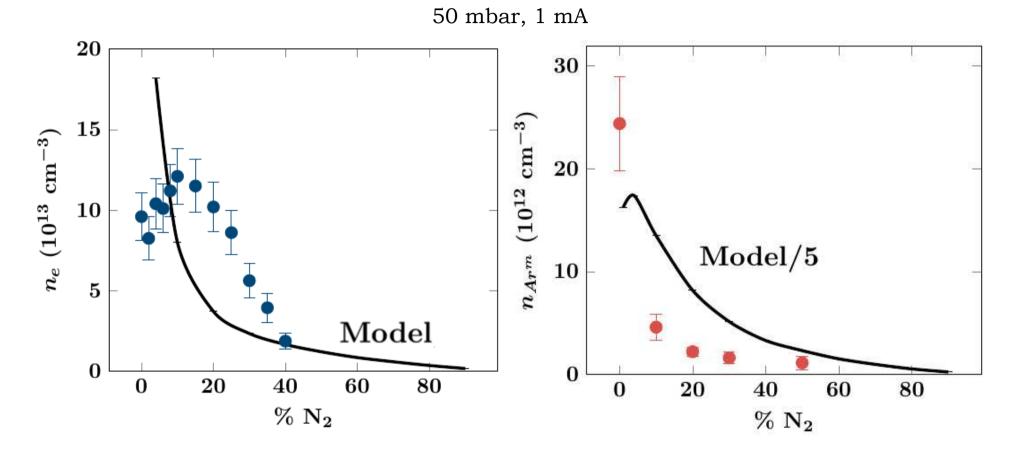
- High density near the cathode surface and then fast decay
- Decrease in agreement with the diffusion model

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- Good agreement between TALIF and FTS; higher calculated densities
- Increase of the density with pressure and % of N_2 ; discrepancy between the two technics at higher pressure
- Opposite trend model/experiments at high % of N_2

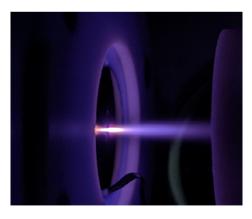
n_N: effect of dilution



Decrease of n_e and n_{Arm} + increase of $N_2^0 \rightarrow$ optimum of % N_2 for n_N ?



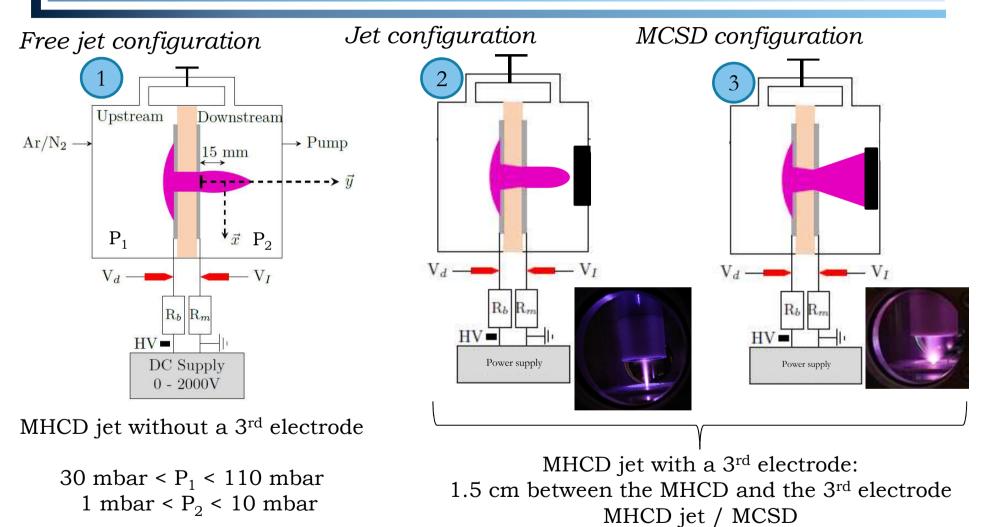
Study of the 1 MHCD reactor: transport of atomic nitrogen





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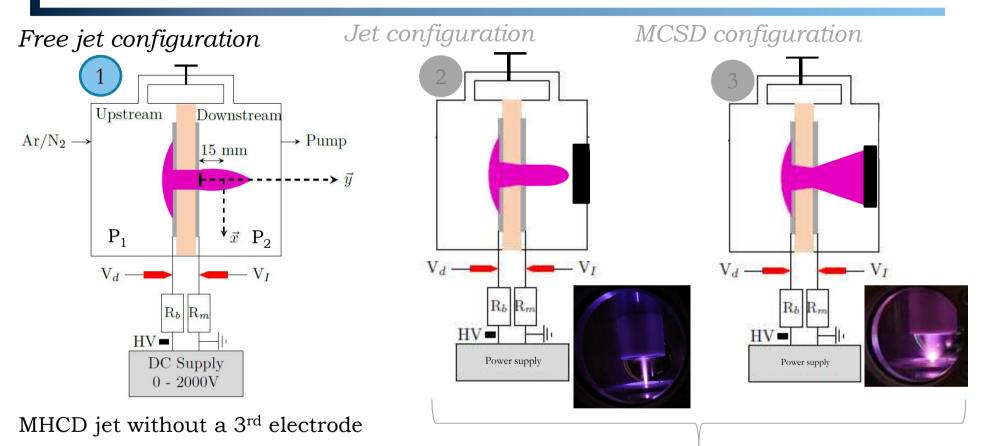
Experimental setups for the MHCD characterization



(Substrate holder not polarized / polarized)



Experimental setups for the MHCD characterization



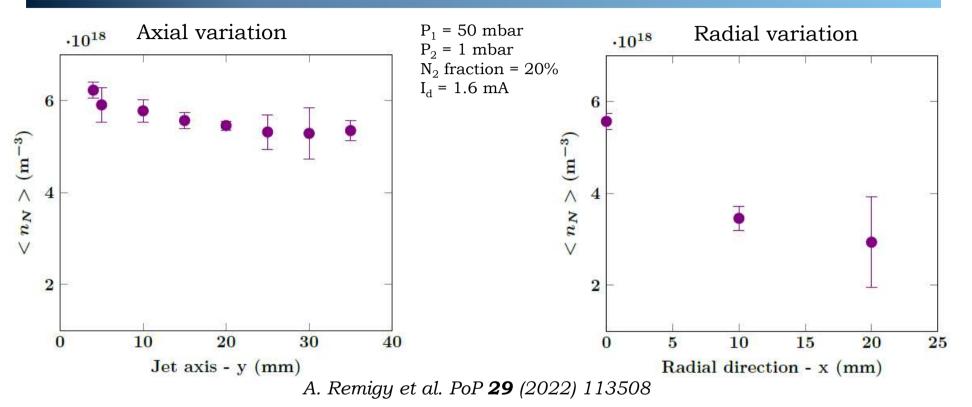
30 mbar < P₁ < 110 mbar

1 mbar < P_2 < 10 mbar

MHCD jet with a 3rd electrode: 1.5 cm between the MHCD and the 3rd electrode MHCD jet / MCSD (Substrate holder not polarized / polarized)



Spatial mapping of N



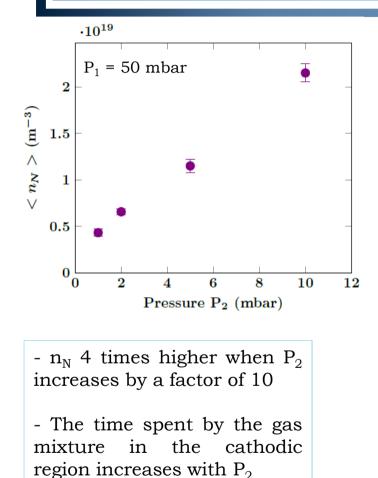
- Density of N atoms almost constant along the jet axis: decrease of 16% over 35 mm

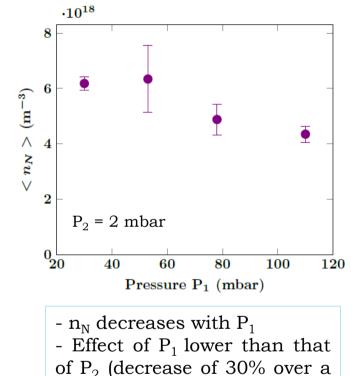
 \rightarrow Production in the cathodic region and transport along the anodic chamber due to the pressure differential

Decrease of the density along the radial direction: density 2 times lower at x=20 mm
→ could be overcome using an array of holes (MHCDs in parallel)



Effect of the pressures

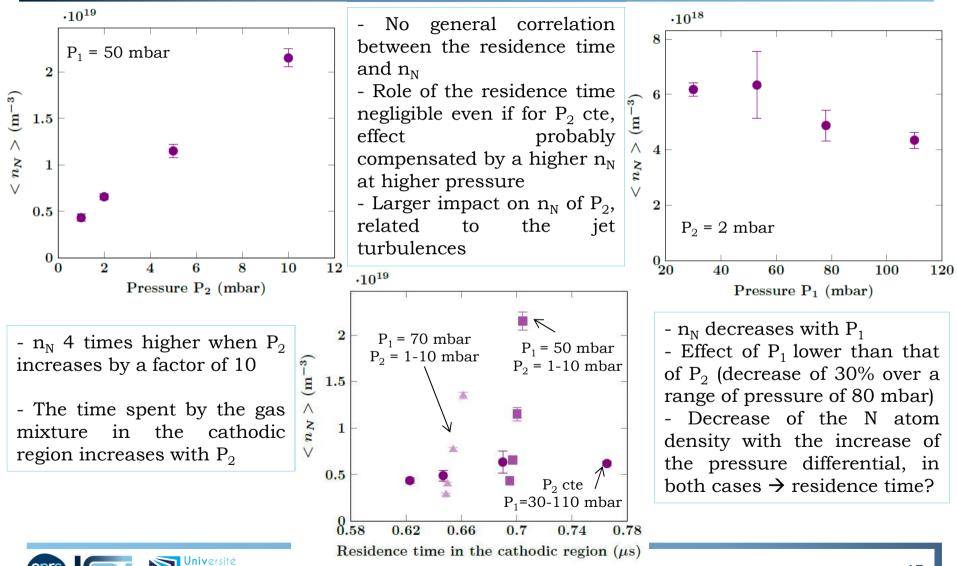




range of pressure of 80 mbar) - Decrease of the N atom density with the increase of the pressure differential, in both cases \rightarrow residence time?



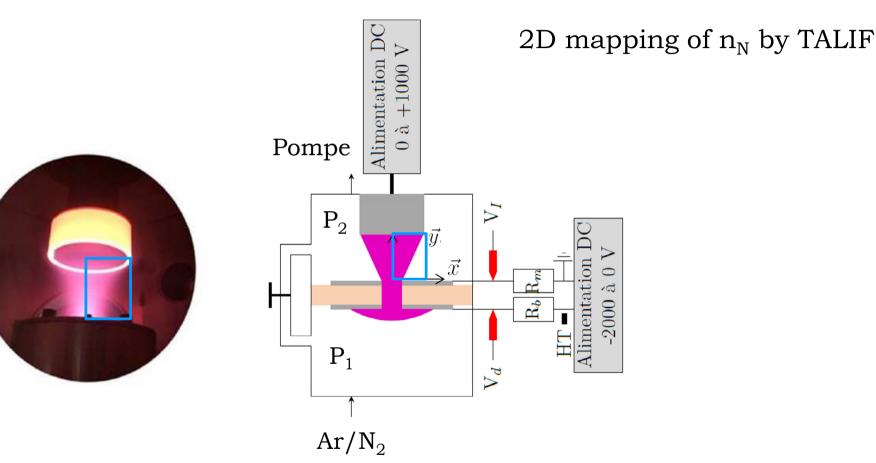
Effect of the pressures





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n_N in the deposition chamber 23

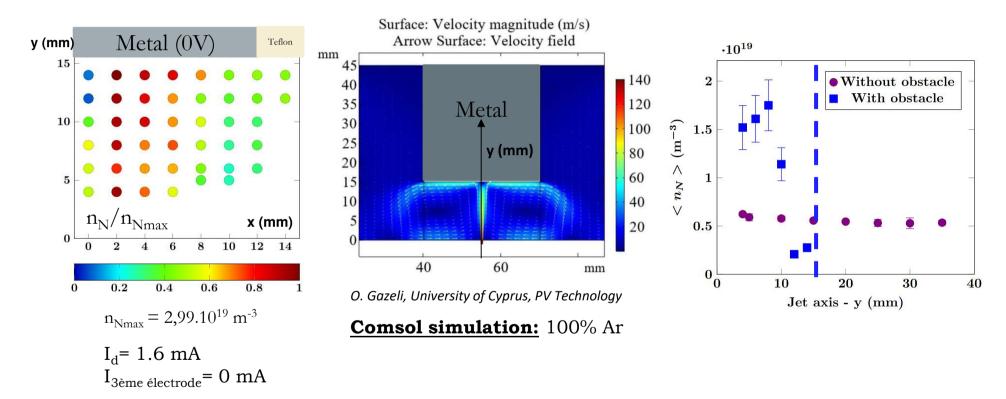


Configurations jet + MCSD



Spatial mapping of N: jet configuration

<u>Conditions</u>: 80% N₂ - 20% Ar, 26 sccm, 30 mbar – 10 mbar

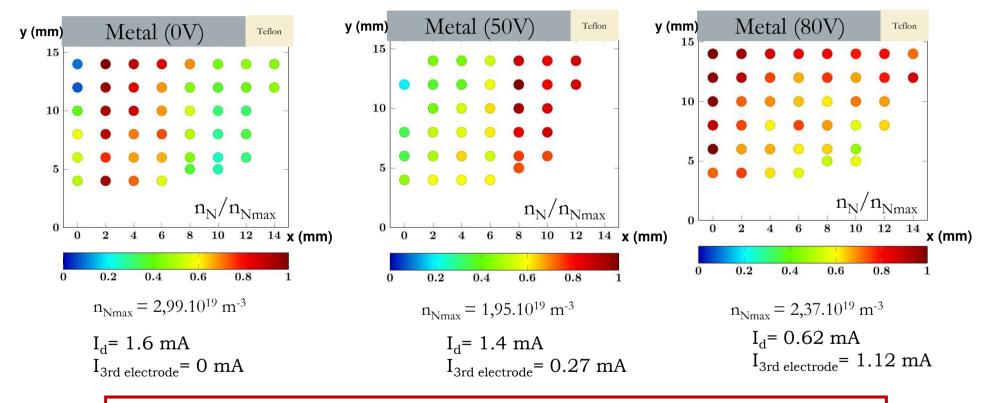


- With an obstacle: depletion in the center \rightarrow gas flow
- In that case we do not have a film deposition

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Spatial mapping of N: MCSD

Conditions: 80% N₂ - 20% Ar, 26 sccm, 30 mbar - 10 mbar



- Mode transition required to homogenize the density
- Almost homogenous density over the whole substrate



3

Conclusion

- MHCD is an effective way to produce high densities of N atoms and cover a high area substrate homogenously → low cost process for h-BN deposition
- Small impact of P_1 on n_N , larger effect of P_2 : great impact of fluid effects, more than the expected effect of the resident time
- Presence of N atoms near the substrate not enough for a deposition : bias needed
- 3rd electrode polarized with a voltage high enough to have more current flowing through the second anode → homogenous density on the substrate + dissociation of boron precursor?



Perspectives

- Future deposition experiments: increase of the pressure in the deposition chamber (P_2) and variation of the bias on the substrate holder
- Improvement of the global model: chemistry of discharge → vibrational levels of N₂, eedf in presence of N₂ → Boltzmann's solver, treatment of the wall losses → analytical model of the sheath (+ comparison to PIC simulations)
- Focus on B atoms (mapping in the deposition chamber) + modelisation of the gas flow → design of the future reactor (new MHCD source, new heating system of the substrate holder)











Thanks for your attention

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Alice Remigy is looking for a post-doc position from August 1st 2023: alice.remigy@cnrs.fr